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Preparation of superconducting Tl-Ba-Ca-Cu-O thin films by Tl2 O3 vapor processing

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[54] PREPARATION OF SUPERCONDUCTING TL-BA-CA-CU-O THIN FILMS BY TL_2O_3 VAPOR PROCESSING

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[*] Notice: The portion of the term of this patent subsequent to Mar. 17, 2009 has been disclaimed.

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Related U.S. Application Data

[63] Continuation of Ser. No. 308,896, Feb. 9, 1989, Pat. No. 4,997,811, which is a continuation-in-part of Ser. No. 236,507, Aug. 25, 1988.

[51] **Int. Cl.**[5] **B05D 5/12**

[52] **U.S. Cl.** **505/1;** 505/742; 427/62; 427/126.3; 427/248.1; 427/343

[58] **Field of Search** 505/1, 731, 732, 783, 505/742; 427/62, 63, 126.3, 248.1, 255, 333, 343, 53.1

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[57] **ABSTRACT**

A process of making high temperature Tl-based superconductors is disclosed. The process includes the steps of reacting solid Ba-Ca-Cu-oxides with Tl_2O_3 vapor. The process allows high quality Tl-based superconductors to be easily fabricated.

11 Claims, 7 Drawing Sheets

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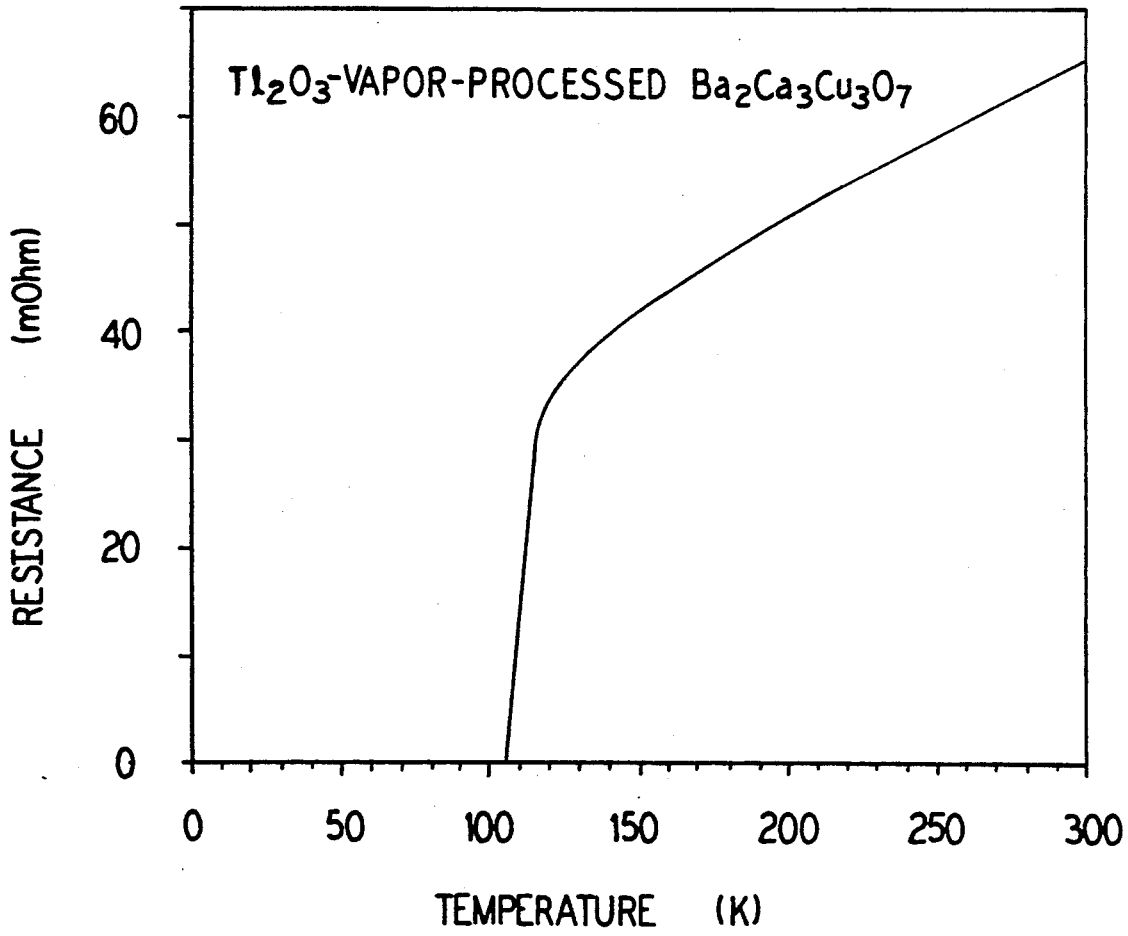
FIG. 1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RESISTANCE (mOhm)
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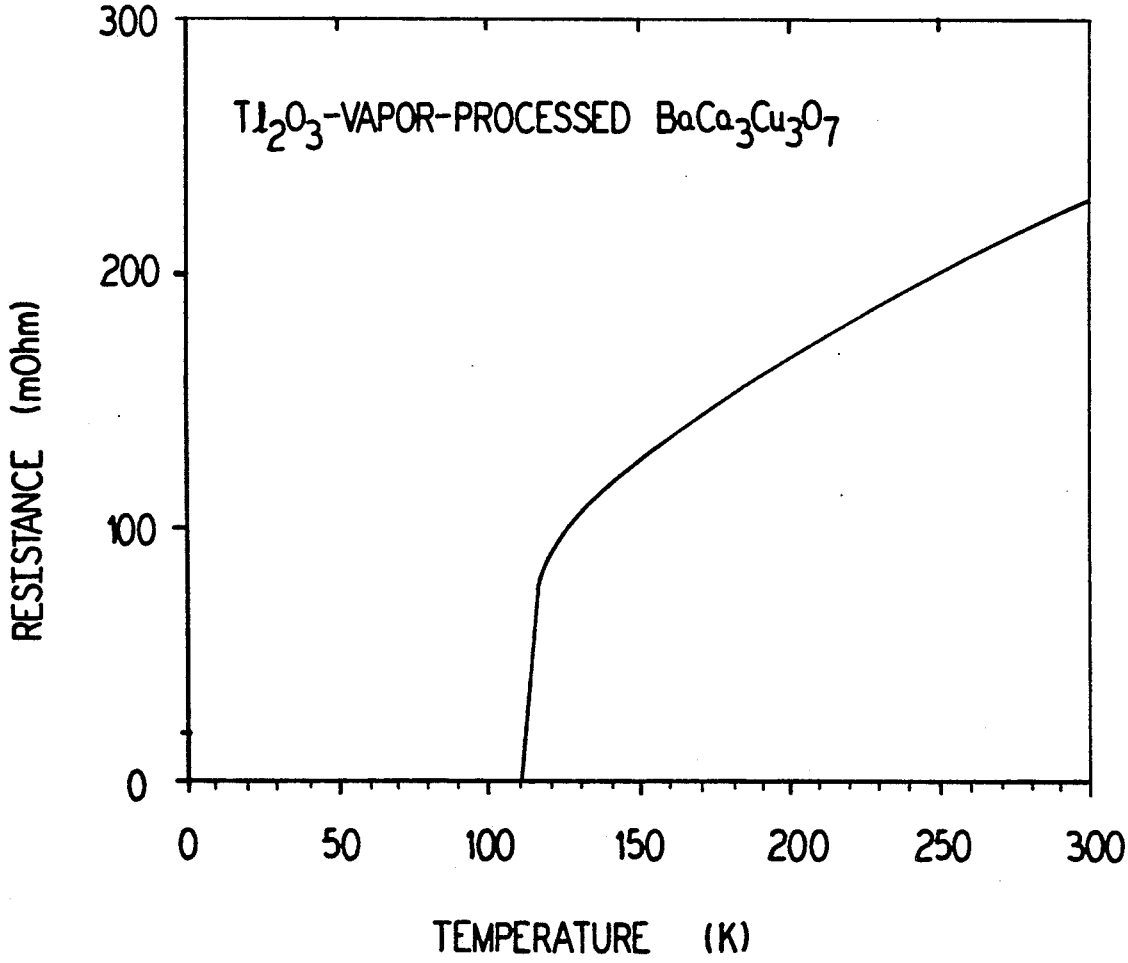
20
40
60
0
50
100
150
200
250
300
TEMPERATURE (K)

FIG. 2


300
200
100
0
Tl_2O_3-VAPOR-PROCESSED $BaCa_3Cu_3O_7$
RESISTANCE (mOhm)
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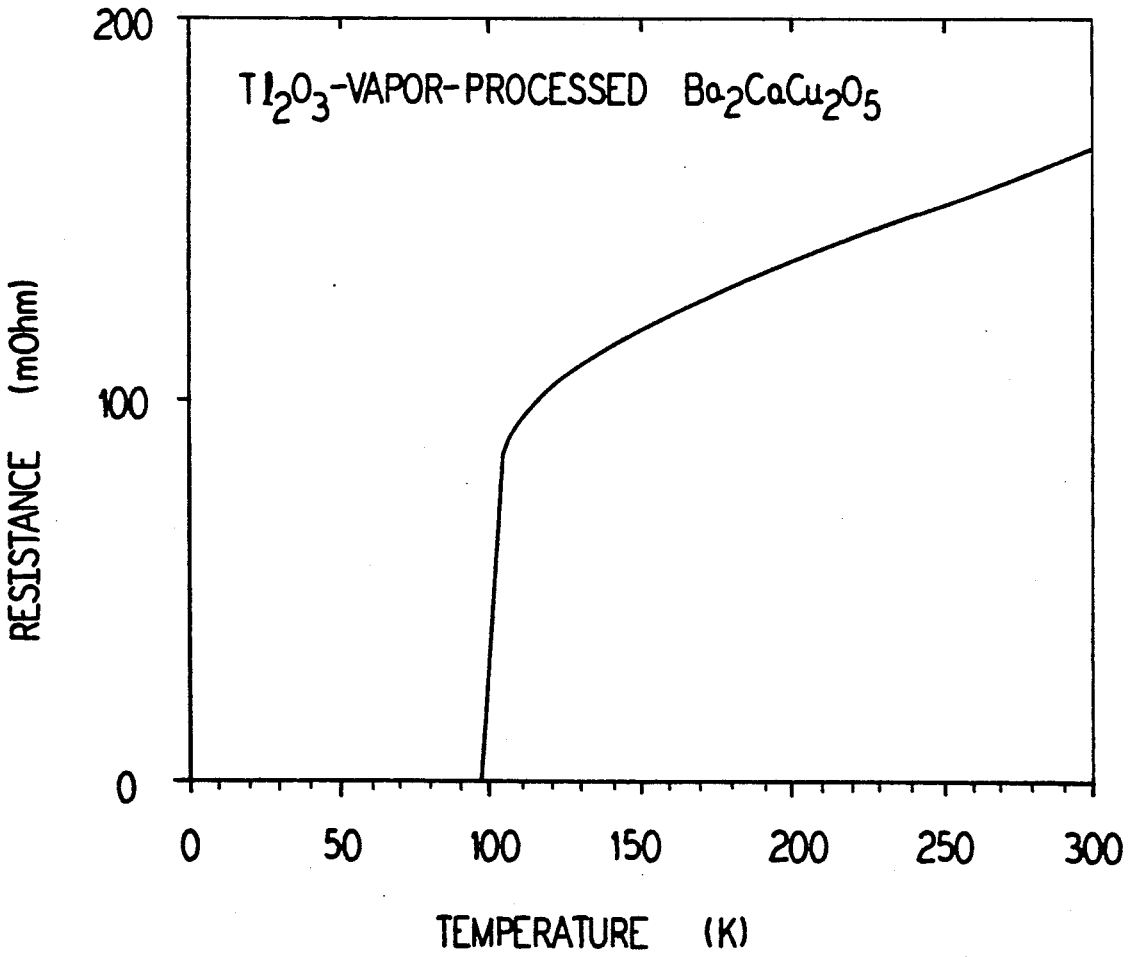
50
100
150
200
250
300
TEMPERATURE (K)

FIG. 3


Tl_2O_3-VAPOR-PROCESSED $Ba_2CaCu_2O_5$
RESISTANCE (mOhm)
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100
0
0
50
100
150
200
250
300
TEMPERATURE (K)

FIG. 4

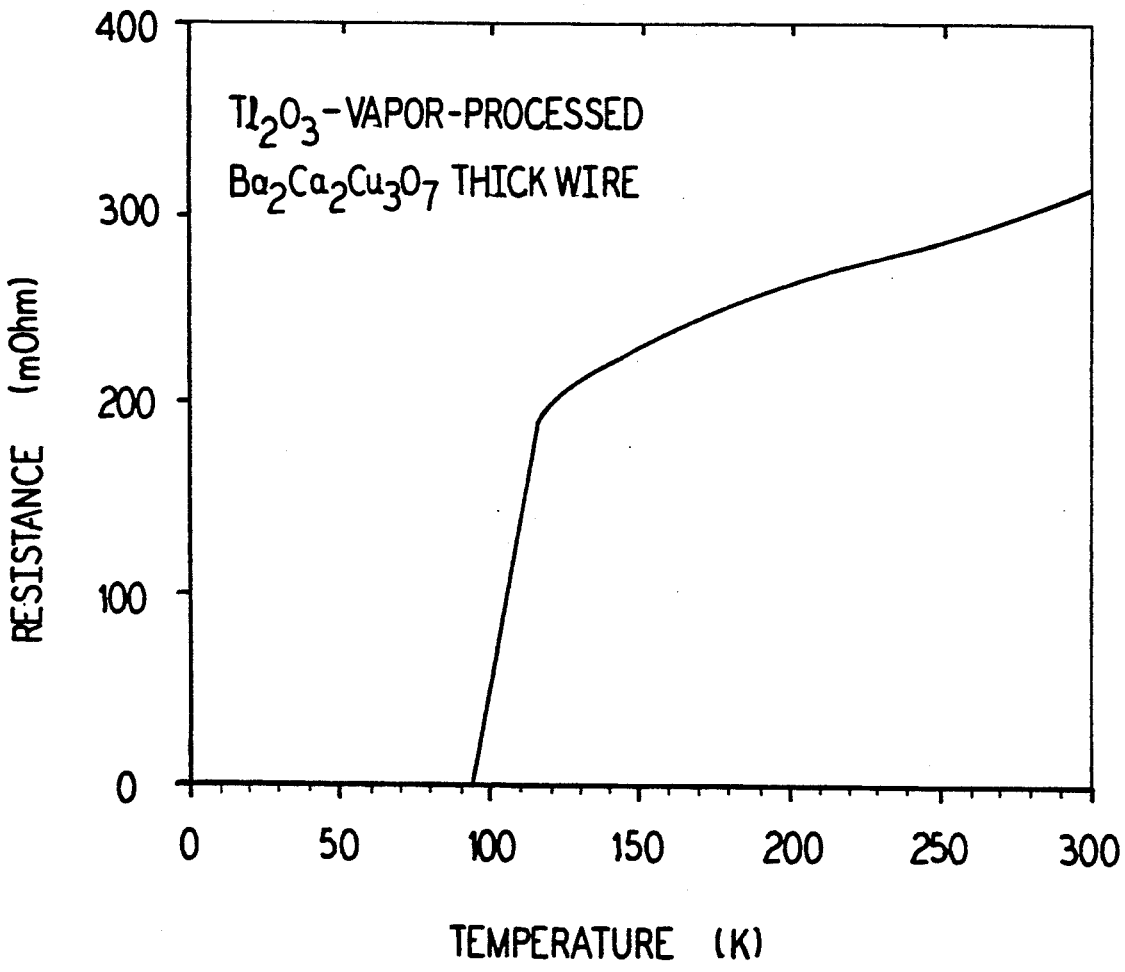

400
300
200
100
0
RESISTANCE (mOhm)
Tl_2O_3-VAPOR-PROCESSED
$Ba_2Ca_2Cu_3O_7$ THICK WIRE
0
50
100
150
200
250
300
TEMPERATURE (K)

FIG. 5

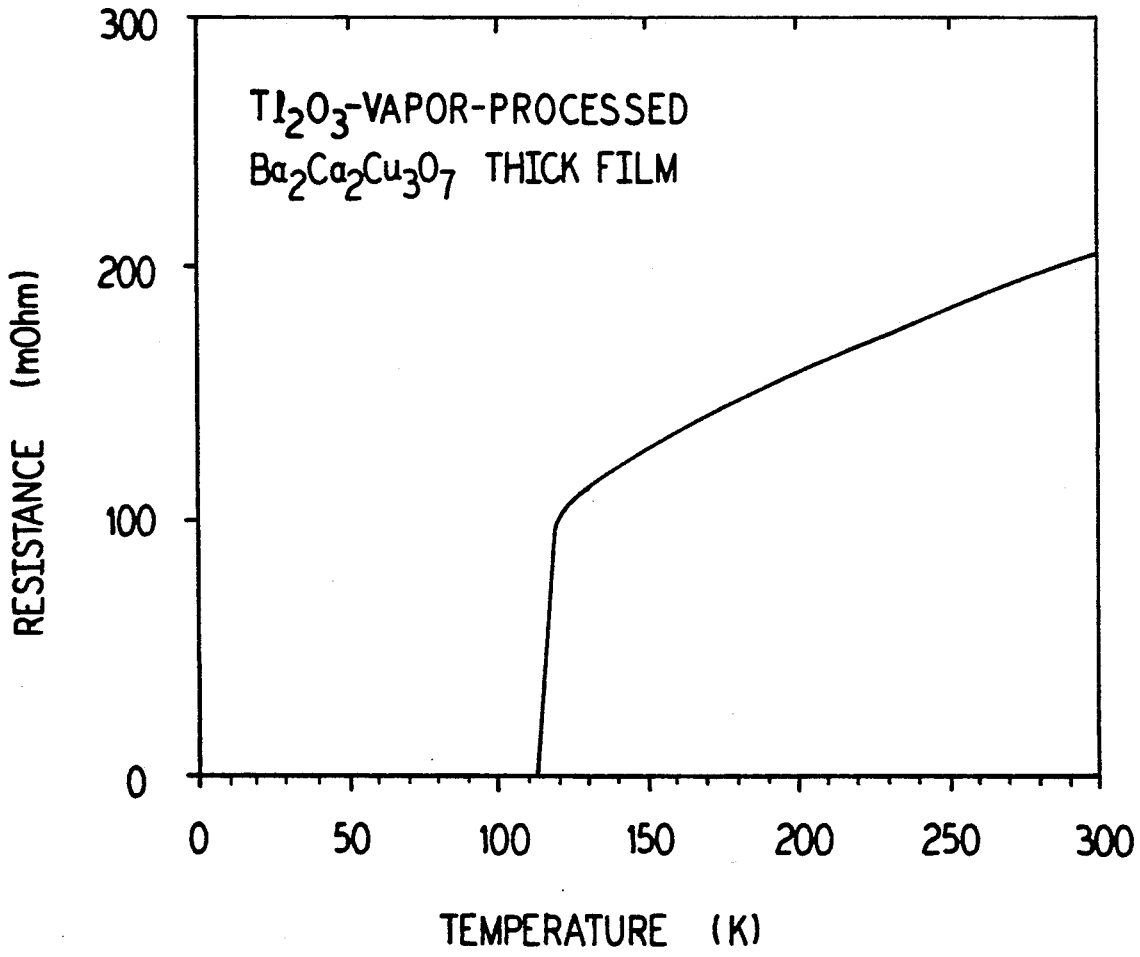

300
200
100
0
RESISTANCE (mOhm)
Tl_2O_3-VAPOR-PROCESSED
$Ba_2Ca_2Cu_3O_7$ THICK FILM
0
50
100
150
200
250
300
TEMPERATURE (K)

FIG. 6

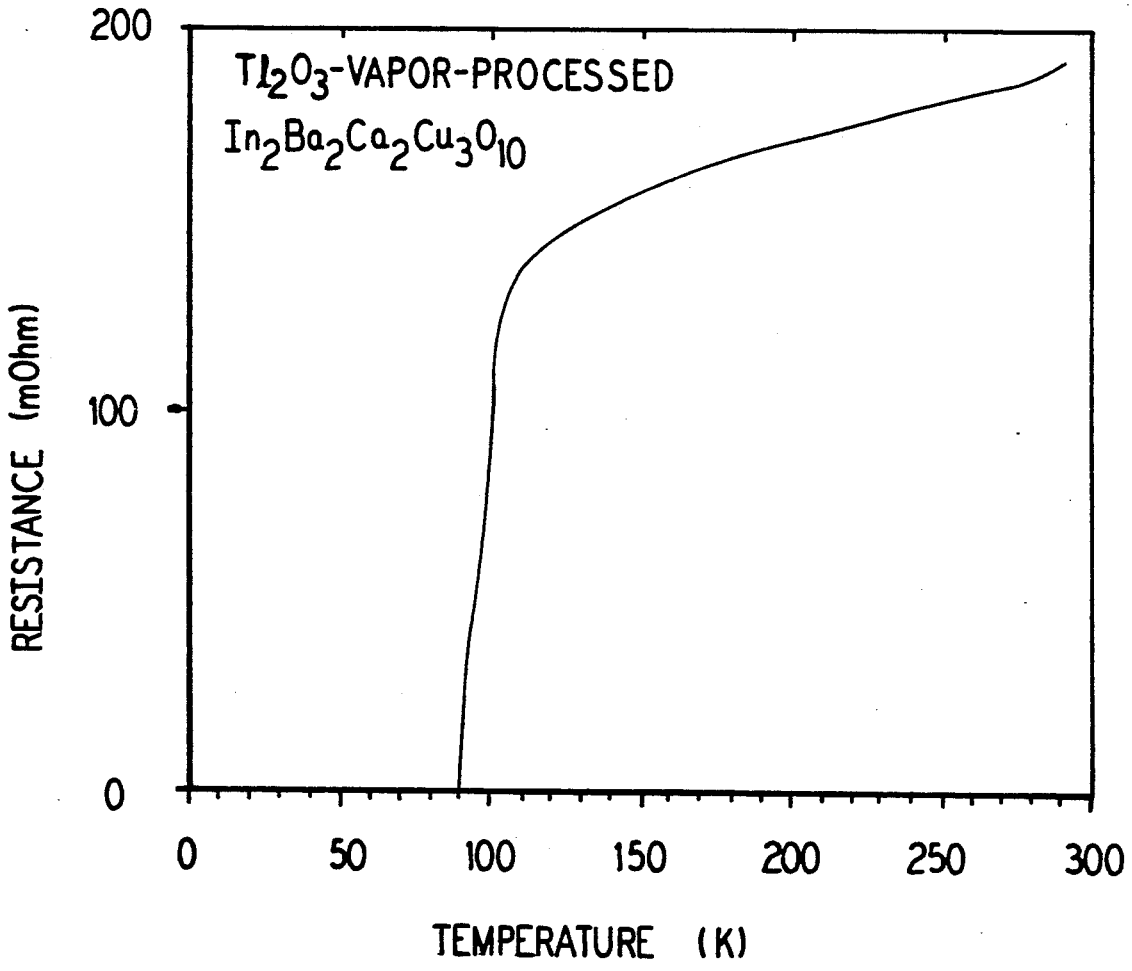
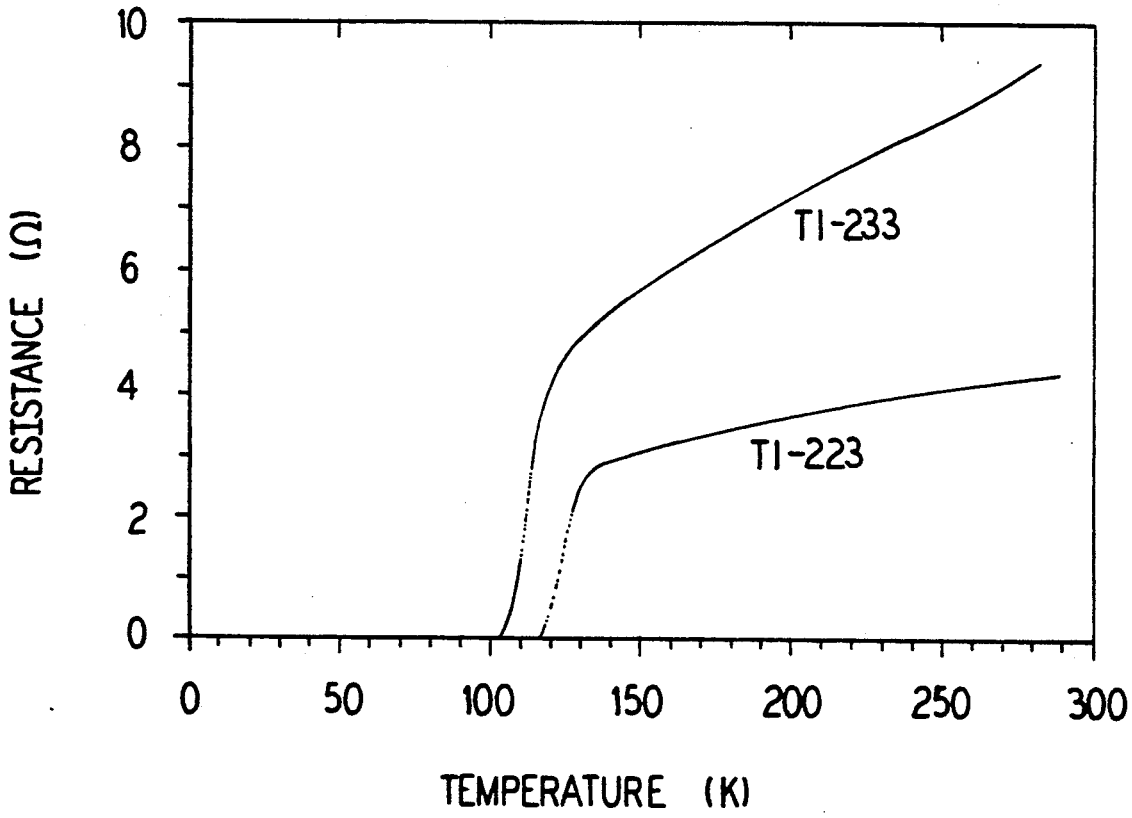

200
100
0
RESISTANCE (mOhm)
Tl_2O_3-VAPOR-PROCESSED
$In_2Ba_2Ca_2Cu_3O_{10}$
0
50
100
150
200
250
300
TEMPERATURE (K)

FIG. 7


10
8
6
4
2
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RESISTANCE (Ω)
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Tl-223
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50
100
150
200
250
300
TEMPERATURE (K)

PREPARATION OF SUPERCONDUCTING TL-BA-CA-CU-O THIN FILMS BY TL_2O_3 VAPOR PROCESSING

This is a continuation of application Ser. No. 308,896 filed Feb. 9, 1989, now U.S. Pat. No. 4,997,811, which is a continuation-in-part of U.S. patent application Ser. No. 236,507 filed on Aug. 25, 1988.

BACKGROUND OF THE INVENTION

The present invention relates generally to high temperature superconductors More specifically, the present invention relates to the fabrication of the high temperature Tl-Ba-Ca-Cu-O superconductors.

U.S. patent application Ser. No. 155,247 now U.S. Pat. No. 4,962,083, filed in the name of the inventors, of the present patent application discloses, in part, Tl-Ba-Ca-Cu-O superconductors that were discovered by the present inventors. These Tl-Ba-Ca-Cu-O superconductors have a transition temperature up to and above 120 K, this transition temperature, the inventors of the present patent application believe, is the highest to date among all existing high temperature superconductors

For useful operation, the transition temperature of a superconductor typically must be at least 1/3 higher than the temperature of operation. For this reason, the Tl-Ba-Ca-Cu-O system has been called the first real liquid nitrogen temperature superconducting system.

The present existing procedures of preparing Tl-based superconductors involve the mixing of all component elements followed by a final sinter. Since Tl_2O_3 evaporates easily, the quality of the Tl-samples is not easily controlled. In addition, in view of its toxicity, Tl_2O_3 presents an additional set of difficulties during the fabrication of these types of superconductors.

An improved method for making Tl-Ba-Ca-Cu-O superconductors would be desirable for many reasons such a process would: (1) allow Tl-based superconductors to be easily constructed ±n the forms of complex bulk components, wires and fibers, and thick and thin films; (2) minimize the toxicity problem caused by Tl compounds; and (3) provide low cost processing and manufacturability.

Accordingly, there is a need for a new process of making Tl-based high temperature superconductors.

SUMMARY OF THE INVENTION

The present invention provides a new process for making Tl-based superconductors The process includes the step of reacting vapor phase Tl_2O_3 and solid Ba-Ca-Cu-oxides. Pursuant to the method of the present invention the Tl-based superconductors are produced in approximately two steps. The first step is the preparation of Ba-Ca-Cu-oxides; and the second step is the processing of the Ba-Ca-Cu-oxides by the use of Tl_2O_3 vapor to form high quality Tl-based superconductors. Pursuant to the present invention, the method of making Tl-based superconductors is simplified and only requires the manufacture of Ba-Ca-Cu-oxides and a final Tl_2O_3 vapor phase treatment.

Accordingly, an advantage of the present invention is that it provides a method which can easily produce Tl-based superconductors.

A further advantage of the present invention is to provide high quality Tl-based superconductors.

A still further advantage of the present invention is that it provides a method which can make Tl-based superconductors in the forms of bulk materials, wires or fibers, thick and thin films.

Furthermore, an advantage of the present invention is that certain elements can be added into the precursor Ba-Ca-Cu-oxides to satisfy specific needs for various applications.

An additional advantage of the present invention is that the Ba-Ca-Cu-oxides can be made in the recrystalline form of a melt.

Still another advantage of the present invention is that Tl_2O_3-vapor-processing can be carried out in closed containers.

Yet another advantage of the present invention is that it provides a method of making Tl-based superconductors which minimizes contamination by Tl compounds.

Additional advantages and features of the present invention are described in and will be apparent from, the detailed description of the presently preferred embodiments and the drawings.

BRIEF DESCRIPTION OF THE DRAWING

FIG. 1 illustrates the relationship of electrical resistance versus temperature for a Tl_2O_3-vapor-processed $Ba_2Ca_2Cu_3O_7$ sample made pursuant to the method of the present invention.

FIG. 2 illustrates the relationship of electrical resistance versus temperature for a Tl_2O_3-vapor-processed $BaCa_3Cu_3O_7$ sample made pursuant to the method of the present invention.

FIG. 3 illustrates the relationship of electrical resistance versus temperature for a Tl_2O_3-vapor-processed $Ba_2CaCu_2O_5$ sample made pursuant to the method of the present invention.

FIG. 4 illustrates resistance-temperature dependence for a Tl_2O_3-vapor-processed $Ba_2Ca_2Cu_3O_7$ thick wire sample made pursuant to the method of the present invention.

FIG. 5 illustrates resistance-temperature dependence for a Tl_2O_3-vapor-processed $Ba_2Ca_2Cu_3O_7$ recrystallized thick film sample made pursuant to the method of the present invention.

FIG. 6 illustrates resistance as a function of temperature for a Tl_2O_3-vapor-processed $In_2Ba_2Ca_2Cu_3O_{10}$ sample made pursuant to the present invention.

FIG. 7 illustrates the relationship of electrical resistance versus temperature for a Tl_2O_3 vapor processed $Ba_2Ca_2Cu_3O_7$ thin film (Tl-223) made pursuant to the present invention and a Tl_2O_3 vapor processed $Ba_2Ca_3Cu_5O_8$ thin film (T1-223) made pursuant to the present invention.

DETAILED DESCRIPTION OF THE PRESENTLY PREFERRED EMBODIMENTS

The present invention provides a method for making high quality Tl-Ba-Ca-Cu-O superconductors. The present invention is based on reactions between vapor phase Tl_2O_3 and solid Ba-Ca-Cu-oxides. The fabrication procedure for the making Tl-Ba-Ca-Cu-O superconductors according to the present invention can be divided into two steps: (1) preparation of Ba-Ca-Cu-oxides; and (2) Tl_2O_3-vapor-processing of the Ba-Ca-Cu-oxides to form Tl-Ba-Ca-Cu-O superconductors The present invention simplifies the fabrication of Tl-Ba-Ca-Cu-oxides to the fabrication of Ba-Ca-Cu-oxides, and minimizes problems caused by the toxicity and volatility of Tl compounds. The present invention allows high quality Tl-based superconductors to be easily made in the forms of complex bulk components, wires and fibers, and

thick and thin films, and provides low cost processing and manufacturability of Tl-based superconductors.

Pursuant to the method of the present invention, first Ba-Ca-Cu-oxides are produced. Preferably compounds, the Ba-Ca-Cu-oxides are produced by grinding and mixing one of the following groups of: $BaCO_3$, CaO, CuO; $BaCO_3$, $CaCO_3$, CuO; BaO_2, CaO_2, CuO; or BaO_2, CaO, CuO. In a preferred embodiment, the molar ratio is 2:2:3, respectively, of the components. In an embodiment, the molar ratio of $BaCO_3$, CaO, CuO is 1:3:3.

Preferably, after the grinding and mixing the powder is heated. Preferably the powder is heated to approximately 925° C. for approximately 24 to about 48 hours with intermediate grindings.

In an embodiment the resultant powder has the following nominal composition:

$$BaCa_xCu_yO_{1+x+y}$$

wherein:

$O \leqq X \leqq 100$; and $O \leqq y \leqq 100$.

In an embodiment, the powder is then pressed into a pellet.

After the powder is prepared, or pressed into a pellet, it can then be reacted with Tl_2O_3 vapor. The Tl_2O_3 vapor can be produced by heating Tl_2O_3 with the Ba-Ca-Cu-oxides or heating any Tl-containing compound that can form Tl_2O_3 vapor upon being heated in oxygen to, for example, approximately 900° C.

In an embodiment, prior to being reacted with Tl_2O_3 vapor, the powder is mixed with a compound chosen from the group consisting of: Tl_2O_3, MgO, AgO, K_2O, Na_2O, ZrO_2, In_2O_3, HgO, Bi_2O_3, PbO_2, and KCl. The resultant mixture is then reacted with Tl_2O_3 vapor.

In an embodiment of the method of the present invention, a Ba-Ca-Cu-oxide film is created and processed with Tl_2O_3 to creating a superconducting thin film.

By way of example and not limitation, examples of the of the present invention process of making high quality Tl-Ca-Ba-Cu-O superconductors will now be given.

EXAMPLE 1

A. The following reagents were utilized:
1. Tl_2O_3,
2. CaO,
3. $BaCO_3$,
4 CuO.

B. The following procedure was followed:
1. A mixture of a two molar portion of $BaCO_3$, a two molar portion of CaO, and a three molar portion of CuO was ground with an agate mortar and pestle, heated in air at 925° C. for more than 24 hours (with several intermediate grindings) to obtain an uniform black $Ba_2Ca_2Cu_3O_7$ powder.
2. The resulting $Ba_2Ca_2Cu_3O_7$ powder was completely ground, and pressed into a pellet.
3. The pellet was heated at about 925° C. in a tube furnace for about 10 minutes.
4. The heated pellet was then taken out of the furnace and cooled in air to room temperature.
5. A small amount (approximately 0.1 to about 0.2 gm) of Tl_2O_3 was put in a platinum boat, and the platinum boat was placed in a quartz boat.
6. The cooled pellet was placed over the platinum boat.
7. The quartz boat, containing the platinum boat, was placed into a tube furnace which had been heated to about 900° C., and was heated for about 3 minutes in flowing oxygen.
8. The sample was then furnace-cooled to room temperature in flowing oxygen, the sample was then taken out of the furnace.

The samples prepared by this procedure formed a layer of superconductive compounds on their bottom surface, which had an onset temperature of above 120 k and a zero resistance temperature of above 100 k. FIG. 1 illustrates the resistance as a function of temperature for a Tl_2O_3-vapor-processed $Ba_2Ca_2Cu_3O_7$ sample made pursuant to this example. This sample reaches zero resistance at 104 k. FIG. 4 illustrates comparable behavior for a Tl_2O_3 vapor-processed $Ba_2Ca_2Cu_3O_7$ thick wire precursor made by a similar procedure.

EXAMPLE 2

A. The following reagents were utilized:
1. Tl_2O_3,
2. CaO,
3. $BaCO_3$,
4. CuO.

B. The following procedure was followed:
1. A mixture of a one molar portion of $BaCO_3$, a three molar portion of CaO, and a three molar portion of CuO was ground with an agate mortar and pestle, heated in air at 925° C. for more than 24 hours (with several intermediate grindings) to obtain a uniform black $BaCa_3Cu_3O_7$ powder.
2. The resulting $BaCa_3Cu_3O_7$ powder was completely ground, and pressed into a pellet.

The pellet was heated at approximately 925° C. in a tube furnace for about 10 minutes.
4. The heated pellet was then taken out of the furnace and cooled in air to room temperature.
5. A small amount (approximately 0.1 to about 0.2 gm) of Tl_2O_3 was placed in a platinum boat, and the platinum boat was then placed in a quartz boat.
6. The cooled pellet was placed over the platinum boat.
7. The quartz boat was put into the tube furnace which had been heated to about 900° C., and was heated for about 3 minutes in flowing oxygen.
8. The sample was then furnace-cooled to room temperature in flowing oxygen, and the sample was then taken out of the furnace.

The samples prepared by this procedure formed a layer of superconductive compounds on their bottom surface, which had an onset temperature of above 120 K, and a zero resistance temperature of above 100 K. FIG. 2 illustrates resistance as a function of temperature for a Tl_2O_3-vapor-processed $BaCa_3Cu_3O_7$ sample prepared pursuant to this example. This sample reaches zero resistance at 110 K. FIG. 3 illustrates comparable behavior for a $Ba_2CaCu_2O_5$ precursor similarly prepared.

EXAMPLE 3

A. The following reagents were utilized:
1. Tl_2O_3,
2. $CaCO_3$,
3. $BaCO_3$,
4 CuO.

B. The following procedure was followed:
1. A mixture of a two molar portion of $BaCO_3$, a two molar portion of $CaCO_3$, and a three molar portion

of CuO was ground with an agate mortar and pestle, heated in air at approximately 925° C. for more than 24 hours (with several intermediate grindings) to obtain a uniform black $Ba_2Ca_2Cu_3O_7$ powder.

2. The resulting $Ba_2Ca_2Cu_3O_7$ powder was completely ground, and pressed into a pellet.
3. The pellet was heated at approximately 925° C. in a tube furnace for about 10 minutes.
4. The heated pellet was then taken out of the furnace and cooled in air to room temperature.
5. A small amount (approximately 0.1 to about 0.2 gm) of Tl_2O_3 was put in a platinum boat, and the platinum boat was placed in a quartz boat.
6. The cooled pellet was placed over the platinum boat.
7. The quartz boat was put into the tube furnace which had been heated to approximately 900° C., and was heated for about 3 minutes in flowing oxygen.
8. The sample was then furnace-cooled to room temperature in flowing oxygen, and was then removed from the furnace.

The samples prepared by this procedure formed a layer of superconductive compounds on their bottom surface, which have an onset temperature of above 120 K, a midpoint of about 110 K, and a zero resistance temperature of above 100 K.

EXAMPLE 4

A. The following reagents were utilized:
1. Tl_2O_3,
2. CaO,
3. $BaCO_3$,
4. CuO.

B. The following procedure was followed:

1. A mixture of a two molar portion of $BaCO_3$, a two molar portion of CaO, and a three molar portion of CuO was ground with an agate mortar and pestle, heated in air at 925° C. for more than 24 hours (with several intermediate grindings) to obtain a uniform black $Ba_2Ca_2Cu_3O_7$ powder.
2. The resulting $Ba_2Ca_2Cu_3$ powder was placed on a platinum substrate which was put in a quartz boat, and was heated in a tube furnace at approximately 950 to about 1000° C. for 3–5 minutes in flowing oxygen.
3. The quartz boat was then taken out of the furnace. The $Ba_2Ca_2Cu_3O_7$ powder had melted completely, forming a layer of recrystalline Ba-Ca-Cu-O.
4. A small amount (approximately 0.1 to about 0.2 gm) of Tl_2O_3 was put in a platinum boat, and the platinum boat was placed in a quartz boat.
5. The platinum substrate was put over the platinum boat with the molten Ba-Ca-Cu-O facing downward.
6. The quartz boat was put into the tube furnace which had been heated to about 900° C., and was heated for about 3 minutes in flowing oxygen.
7. The sample was then furnace-cooled to room temperature in flowing oxygen, and was then taken out of the furnace.

A Tl_2O_3-vapor-processed $Ba_2Cu_3O_7$ thick film was produced that was superconducting. FIG. **5** illustrates resistance as a function of temperature for a Tl_2O_3-vapor-processed $Ba_2Ca_2Cu_3O_7$ recrystallized thick film created pursuant to this example. The film had an onset temperature of above 120 K and reached zero resistance at about 111 K.

This experiment also shows that thin film Tl-Ca-Ba-Cu-O superconductors can be made using the Tl_2O_3 vapor process with appropriately deposited Ca-Ba-Cu-O precursor thin films. These thin films can be produced by depositing a thin-film of Ca-Ba-Cu-O precursor utilizing known techniques of physical or chemical deposition. These techniques include, inter alia, sputtering, evaporation, ablation, electrodeposition, electroless deposition, and chemical vapor deposition. After the thin film of Ca-Ba-Cu-O is produced, it can then be reacted with Tl_2O_3 vapor, for example, by placing Tl_2O_3 in a boat under or near the precursor and heating the boat.

EXAMPLE 5

A. The following reagents were utilized:
1. Tl_2O_3,
2. In_2O_3,
3. CaO,
4. $BaCO_3$,
5. CuO.

B. The following procedure was followed:

1. A mixture of two molar portion of $BaCO_3$, a two molar portion of CaO, and a three molar portion of CuO was ground with an agate mortar and pestle, heated in air at approximately 925° C. for more than 24 hours (with several intermediate grindings) to obtain a uniform black $Ba_2Ca_2Cu_3O_7$ powder.
2. A one molar portion of the resulting $Ba_2Ca_2Cu_3O_7$ powder was mixed with a one molar portion of In_2O_3, and was completely ground and pressed into a pellet.
3. The pellet was heated in a tube furnace at about 900° C. in flowing oxygen for about 3 minutes, and was then removed from the furnace.
4. A small amount (approximately 0.1 to about 0.2 gm) of Tl_2O_3 was put in a platinum boat, and the platinum boat was put in a quartz boat
5. The $In_2Ba_2Ca_2Cu_3O_7$ pellet was put over the platinum boat.
6. The quartz boat was then put into the tube furnace which had been heated to approximately 900° C., and was heated for about 3 minutes in flowing oxygen.
7. The sample was then furnace-cooled to room temperature in flowing oxygen, and was then removed from the furnace.

The bottom surface of the Tl_2O_3-vapor-processed $In_2Ba_2Ca_2Cu_3O_7$ sample constructed pursuant to this method was found to be superconducting FIG. **6** illustrates resistance as a function of temperature for a Tl_2O_3-vapor-processed $In_2Ba_2Ca_2Cu_3O_7$ sample, made pursuant to this example, which had an onset temperature about 120 K, and reached zero resistance at 89 K.

EXAMPLE 6

A. The following reagents were utilized:
1. Tl_2O_3,
2. CaO,
3. $BaCO_3$,
4. CuO.

B. The following procedure was followed:

A mixture of a two molar portion of $BaCO_3$, a two molar portion of CaO, and a three molar portion of CuO was ground with an mortar and pestle, heated in air at 925° C. for more than 24 hours (with several intermediate grindings) to obtain a uniform black $Ba_2Ca_2Bu_3O_7$ powder.

2. The resulting $Ba_2Ca_2Cu_3O_7$ powder was completely ground, and pressed into a pellet.
3. The pellet was heated in a tube furnace at approximately 925° C. for about 5 minutes.
4. The pellet was then taken out of the furnace and cooled in air to room temperature.
5. A small amount (approximately 0.1 to about 0.2 gm) of Tl_2O_3 was put in a platinum boat, and the platinum boat and cooled pellet were put in a gold container.
6. The gold container, was sealed so that the platinum boat and pellet were sealed in oxygen, was put into a tube furnace which had been heated to approximately 900° C., and was heated for about 10 minutes.
7. The gold container was then furnace-cooled to room temperature, and was removed from the furnace.
8. The gold container was opened, and the sample taken out.

This example produced a Tl_2O_3-vapor-processed sample that formed a layer of superconducting compounds on its surface, which had an onset temperature of above 120 K and a zero resistance temperature of above 100 K.

EXAMPLE 7

A. The following reagents were utilized:
1. Tl_2O_3,
2. CaO,
3. $BaCO_3$,
4. CuO.

B. The following procedure was followed:

A mixture of a two molar portion of $BaCO_3$, a two molar portion of CaO, and a three molar portion of CuO was ground with an agate mortar and pestle, heated in air at approximately 925° C. (with several intermediate grindings) to obtain a uniform black $Ba_2Ca_2Cu_3O_7$ powder.

2. The resulting $Ba_2Ca_2Cu_3O_7$ powder was mixed with Tl_2O_3 in a molar ratio of 1:1, and was completely ground and pressed into a pellet.
3 The pellet was heated at approximately 925° C. in a tube furnace for approximately 4 hours. The sample could have been heated for a longer time however.
4. The heated pellet was then taken out of the furnace and cooled in air to room temperature. The resultant pellet was found to have a semiconductor behavior.
5. A small amount (approximately 0.1 to about 0.2 gm) of Tl_2O_3 was put in a platinum boat, and the platinum boat was put in a quartz boat.
6. The resultant pellet was put over the platinum boat.
7. The quartz boat was then put into the tube furnace which had been heated to approximately 900° C., and was heated for about 3 minutes in flowing oxygen.
8. The sample was then furnace-cooled to room temperature in flowing oxygen, and was then removed from the furnace.

The resulting Tl_2O_3-vapor-processed sample, produced by this example, had a superconducting behavior, and had an onset temperature of above 120 K, and a zero resistance temperature of above 100 K.

EXAMPLE 8

A. The following reagents were utilized:
1. Tl_2O_3,
2. CaO,
3. $BaCO_3$,
4. CuO.

B. The following procedure was followed:

1. A mixture of a two molar portion of $BaCO_3$, a two molar portion of CaO, and a three molar portion of CuO was ground with an agate mortar and pestle, heated in air at 925° C. for more than 24 hours (with several intermediate grindings) to obtain a uniform black $Ba_2Ca_2Cu_3O_7$ powder. A pellet was then formed in a $\frac{3}{4}$ inch die under pressure.
2. A $Ba_2Ca_2Cu_3O_7$ precursor film was then deposited by laser deposition from the pellet as a target onto a Y-stabilized ZrO_2 substrate using a frequency-doubled Nd-YAG laser operating at 523 nm, forming a Ba-Ca-Cu-O thin film with a thickness of approximately 2 to about 3 μm.
3. The thin film precursor was then placed above a platinum boat which contained approximately 0.1 to about 0.2 gram of Tl_2O_3, and the platinum boat was placed in a quartz boat.
4. The quartz boat and its contents was heated at 900 ° C. in a tube furnace for approximately 2 to about 3 minutes in flowing oxygen, and then furnace cooled to room temperature.

A resultant Tl_2O_3 vapor processed $Ba_2Ca_2Cu_3O_7$ thin film was found to be superconducting with an onset temperature above 120 K and a zero resistance temperature of 115 K. The measurements of a sample prepared pursuant to this example are illustrated in FIG. 7 as Tl-223.

EXAMPLE 9

A. The following reagents were utilized:
1. Tl_2O_3,
2. CaO,
3. $BaCO_3$,
4. CuO.

B. The following procedure was followed:

1. A mixture of a two molar portion of $BaCO_3$, a three molar portion of CaO, and a three molar portion of CuO was ground with an agate mortar and pestle, heated in air at 925° C. for more than 24 hours (with several intermediate grindings) to obtain an uniform black $Ba_2Ca_3Cu_3O_8$ powder. A pellet was then formed in a $\frac{3}{4}$ inch die under pressure.
2. A $Ba_2Ca_3Cu_3O_8$ precursor was deposited by laser deposition onto a Y-stabilized ZrO_2 substrate using a frequency-doubled Nd-YAG laser operating at 523 nm, forming a Ba-Ca-Cu-O thin film with a thickness of approximately 2 to about 3 μm.
3. The thin film precursor was placed above a platinum boat which contained approximately 0.1 to about 0.2 gram of Tl_2O_3, and the platinum boat was placed in a quartz boat.
4. The quartz boat and its contents was heated at 900 ° C. in a tube furnace for approximately 2 to about 3 minutes in flowing oxygen, and then furnace cooled to room temperature.

A resultant Tl_2O_3 vapor processed $Ba_2Ca_3Cu_3O_8$ thin film was found to be superconducting with an onset temperature above 120 K and a zero resistance temperature of 104 K. The measurements of a sample prepared

pursuant to this method are illustrated in FIG. 7 as Tl-233.

It should be understood that various changes and modifications to the presently preferred embodiments described here in be apparent to those skilled in the art. Such changes and modifications can be made without departing from the spirit and scope of the present invention and without diminishing its attendant advantages. It is therefore intended that such changes and modifications be covered by the appended claims.

We claim:

1. A method for making thin-film Tl-Ca-Ba-Cu-O superconductors comprising the steps of:

a. depositing a thin-film of Ca-Ba-Cu-O precursor utilizing physical vapor deposition;

b. placing Tl_2O_3 in a boat under or near the precursor thin film;

c. heating the boat and precursor film in flowing O_2; and

d. cooling the film and boat.

2. A method for making thin-film Tl-Ca-Ba-Cu-O superconductors comprising the steps of:

mixing and heating $BaCO_3$, CaO, and CuO to obtain a Ca-Ba-Cu-O powder;

forming a Ca-Ba-Cu-O thin film from the Ca-Ba-Cu-O powder by physical or chemical deposition; and

heating the film in oxygen with a quantity of Tl_2O_3 in oxygen to create a Tl_2O_3 vapor processed film.

3. The method of claim 2 wherein the powder includes the composition $Ba_2Ca_2Cu_3O_7$.

4. The method of claim 2 wherein the powder includes the composition $Ba_2Ca_3Cu_3O_8$.

5. The method of claim 2 wherein two molar portions of $BaCO_3$, two molar portions of CaO, and three molar portions of CuO are ground, mixed, and heated to create the powder.

6. The method of claim 2 wherein two molar portions of $BaCO_3$, three molar portions of CaO, and three molar portions of CuO are ground, mixed, and heated to create the powder.

7. The method of claim 2 wherein the powder is created into a pellet.

8. The method of claim 7 wherein a film is deposited by laser deposition from the pallet into an Y-stabilized ZrO_2 substrate.

9. The method of claim 8 wherein the film has a thickness of approximately 2 to about 3 μm.

10. The method of claim 2 wherein the Tl_2O_3 is placed in a platinum boat and the film is placed above the boat prior to heating the film and Tl_2O_3.

11. The method of claim 2 wherein the quantity of Tl_2O_3 is approximately 0.1 to about 0.2 g.

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